Novel [3 + 2] Cycloaddition Reaction of Alkenyl Fischer Carbene Complexes with Imines Leading to 3-Pyrroline Derivatives

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Fischer carbene complexes have attracted much attention of synthetic chemists because of their diverse reactivity toward unsaturated bonds.¹ Indeed, they have been widely used for cycloaddition reactions with olefins, giving cyclopropanes.^{1b,2} On the other hand, alkenyl Fischer carbene complexes, which are attractive substrates as three-carbon synthons, reacted with both alkenes and dienes to afford cyclopentenes³ and cycloheptadienes,^{1a,4} respectively. Fischer carbene complexes also participated in the synthesis of aza heterocycles,^{5–7} wherein imines have been often used as nitrogen sources.^{5,6d,e,g-j,7} As part of our continued interest in the reaction of imines,⁸ we have developed a novel [3 + 2] cycloaddition reaction of alkenyl Fischer carbene complexes with simple imines. In this communication, we would like to present that alkenyl Fischer chromium carbene complexes react with

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Table 1. Effect of Additives in the [3 + 2] CycloadditionReaction of Alkenyl Carbene Complex **2a** with Imine **1a**^a

entry	additive	yield (%)	trans:cis ^b
1	none	55	84:16
2	ZnCl ₂	60	91:9
3	BF ₃ •OEt ₂	65	80:20
4	GaCl ₃	72	78:22
5^c	GaCl ₃	82	81:19
6	TMSOTf	69	78:22
7	$Sn(OTf)_2$	68	80:20

^{*a*} Molar ratio; **1a:2a**:additive = 1.0:1.0:0.1. ^{*b*} Determined by ¹H NMR. ^{*c*} Molar ratio; **1a:2a**:additive = 1.2:1.0:0.2.

various kinds of simple imines to produce 3-pyrroline derivatives in the presence of a catalytic amount of GaCl₃. This is the first example of the reaction of alkenyl Fischer carbene complex with imines leading to 3-pyrroline derivatives, which are important intermediate for the synthesis of natural and biologically active compounds.⁹

First, the reaction of alkenyl Fischer carbene complex **2a** ($\mathbb{R}^3 = \mathbb{M}e$, $\mathbb{R}^4 = \mathbb{P}h$) with imine **1a** ($\mathbb{R}^1 = \mathbb{R}^2 = \mathbb{P}h$) was chosen as a model reaction (Scheme 1). When a mixture of **1a** (1.0 equiv)

Scheme 1



and 2a (1.0 equiv) was allowed to reflux in 1,2-dichloroethane for 1 h, the corresponding cycloadduct 3a was obtained in 55% yield as a mixture of diastereomers (84:16, determined by 400 MHz ¹H NMR, Table 1, entry 1). The regioselectivity was quite high, and no regioisomeric product was detected. It should be noted that the regioselectivity in the present reaction is quite different from that of the [4 + 3] cycloaddition reactions of alkenyl carbene complexes with 1-amino-1,3-dienes;^{7a,b} the new C-N bond was formed between the β -carbon in the alkenyl carbene complex and the nitrogen in the imine. This fact rules out the initial attack of the nitrogen in the imine to the carbene carbon as proposed in the several reactions of Fischer carbene complexes with nitrogen-containing compounds.7b,d,10 The stereochemistry of the major diastereomer of **3a** was unambiguously determined to be trans by an X-ray crystallographic analysis (see Supporting Information).

Next, to improve the yield, we have examined the effect of additives. Although transition metal compounds (10 mol %) such as RuCl₂(PPh₃)₃ (6%), [Rh(OAc)₂]₂ (20%), PdCl₂(PhCN)₂ (0%), and Cu(OTf)₂ (49%) were not effective as additives, addition of some Lewis acids greatly improved the yields (Table 1, entries 2–7). Among the Lewis acids examined, GaCl₃ (10 mol %) was the most effective; the yield was improved while maintaining the good diastereoselectivity (72%, trans:cis = 78:22, entry 4). TMSOTf and Sn(OTf)₂ were equally effective (entries 6 and 7). The highest yield was obtained when slightly excess amount of **1a** (1.2 equiv) and 20 mol % GaCl₃ were employed (82%, trans: cis = 81:19, entry 5).¹¹

The [3 + 2] cycloaddition reaction of a variety of alkenyl Fischer carbene complexes with imines was carried out under

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⁽¹¹⁾ When the corresponding tungsten complex was allowed to react with **1a** under the same conditions, no cycloadduct was obtained.

Table 2.	[3 +	2]	Cycloaddition	Reactions of	f Alkenyl	Carbene	Complexes	2a-e with	Imines	1a-h	in the	Presence	of	GaC	'l ₃
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entry	imine 1	\mathbb{R}^1	\mathbb{R}^2	complex 2	R ³	\mathbb{R}^4	time (h)	product	yield (%)	trans:cis ^b
1	1a	Ph	Ph	2a	Me	Ph	1	3 a	82	81:19
2	1a	Ph	Ph	2b	Bn	Ph	1	3b	81	85:15
3	1a	Ph	Ph	2c	<i>i</i> -Pr	Ph	1.5	3c	80	86:14
4^c	1a	Ph	Ph	2c	<i>i</i> -Pr	Ph	2	3c	78	90:10
5	1a	Ph	Ph	2d	<i>i</i> -Pr	2-furyl	2	3d	77	81:19
6	1a	Ph	Ph	2e	<i>i</i> -Pr	2-styryl	2.5	3e	66	78:22
7	1b	p-MeC ₆ H ₄	Ph	2c	<i>i</i> -Pr	Ph	1.5	3f	81	87:13
8	1c	<i>p</i> -MeOC ₆ H ₄	Ph	2c	<i>i</i> -Pr	Ph	1	3g	86	83:17
9	1d	p-ClC ₆ H ₄	Ph	2c	<i>i</i> -Pr	Ph	2	3h	64	78:22
10	1e	Ph	p-MeC ₆ H ₄	2c	<i>i</i> -Pr	Ph	1.5	3i	81	89:11
11	1f	Ph	p-MeOC ₆ H ₄	2c	<i>i</i> -Pr	Ph	1	3j	84	92:8
12	1g	Ph	p-ClC ₆ H ₄	2c	<i>i</i> -Pr	Ph	2	3k	65	85:15
13	1ĥ	Bn	Ph	2c	<i>i</i> -Pr	Ph	1.5	31	70	79:21
14 ^c	1h	Bn	Ph	2c	<i>i</i> -Pr	Ph	1.5	31	70	80:20

^{*a*} Molar ratio; **1**:**2**:GaCl₃ = 1.2:1.0:0.2. ^{*b*} Determined by ¹H NMR. ^{*c*} One-pot reaction.

the optimized conditions described above (Scheme 1, Table 2).¹² It was found that bulkiness of the alkoxy substituents (OR³) slightly affected the diastereoselectivities in the reactions; bulkier alkoxy substituents exhibited the better selectivities (entries 1-3). Isopropoxy substituted complexes 2c-e, being readily accessible, were used for further reactions. Replacement of the phenyl group on the β -carbon in the complex with either 2-styryl or 2-furyl groups also worked well to furnish the corresponding adducts 3d and 3e (entries 5 and 6). A variety of aromatic imines, derived from both *p*-substituted benzaldehydes and *p*-substituted anilines, could be used for the present [3 + 2] cycloaddition reactions. In all cases, the corresponding 3-pyrroline derivatives were obtained in high yields and with good diastereoselectivities (entries 7-12). In addition, the [3 + 2] cycloaddition reaction of N-benzyl imine 1h was also found to be successful (entry 13).

Since it is synthetically convenient if imines were prepared in situ and were allowed to react with the complex under one-pot conditions, we examined to apply these reactions to one-pot, threecomponent synthesis of 3-pyrroline derivatives. As a result, it was found that the preparation of imines from benzaldehyde and aniline or benzylamine in the presence of MS4A in 1,2dichloroethane followed by treatment of 2c with 20 mol % GaCl₃ afforded the 3-pyrroline derivatives 3c (entry 4) and 3l (entry 14), respectively.

The synthetic utility of *trans*-3c is exemplified as below (Scheme 2). Hydrolysis of trans-3c with 6 M HCl in THF afforded

Scheme 2



3-pyrrolidinone trans-4 without any epimerization. In addition, trans-3c could be readily converted into the corresponding 3-isopropoxypyrrole derivative 5 by oxidation with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ).¹³ Although formation of pyrrole derivatives by use of Fischer carbene complexes has been described in several literatures, 6a-f,h-l the present [3 + 2] cycloaddition reaction should provide a useful alternative route for synthesis of these compounds as well as 3-pyrroline derivatives.

At present, reaction mechanism including the role of Lewis acid is not clear. One of the tentatively proposed mechanisms is

Scheme 3



outlined in Scheme $3.^{14}$ The [2 + 2] cycloaddition reaction between the Cr-C double bond of the complex 2 and the C-N double bond of the imine 1 leads to two diastereomeric fourmembered chromacycles 6a and 6b.15 The fact that increasing the size of the alkoxy group at the carbone carbon improved the diastereoselectivity implies that **6a** would be formed preferentially over **6b**. Fomation of a six-membered chromacycle **8** via the η^3 allyl complex 7,36,16 followed by reductive elimination leads to trans-3.

In conclusion, we have found a novel [3 + 2] cycloaddition reaction of alkenyl Fischer carbene complexes with simple imines, a reaction which afforded 3-alkoxy-2,5-disubstituted-3-pyrroline derivatives in high yields and with good diastereoselectivities. We are currently investigating the substrate scope, the mechanism including the role of Lewis acid, and the development of an asymmetric version of this process.

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Supporting Information Available: Experimental procedures, characterization data for compounds 2b-e, 3a-l, 4, and 5, and X-ray structural information on trans-3a (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹²⁾ For all cycloadducts, except for 3d and 3e, the relative configurations were assigned on the basis of physical analogy with 3a. The coupling constants $(J(H^2-H^5))$ of the trans diastereomer of **3a** are 5.6 Hz, whereas those of the cis diastereomer of 3a show smaller values of 2.1 Hz. In addition, the trans diastereomer is less polar than the cis isomer. The relative configurations of 3d and 3e were tentatively assigned since their physical properties were shown

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